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Key indicators

Single-crystal X-ray study T = 297 K Mean σ (C–C) = 0.003 Å R factor = 0.042 wR factor = 0.120 Data-to-parameter ratio = 9.0

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

seco-Dukunolide F: a new tetranortriterpenoid from Lansium domesticum Corr.

The title compound, methyl 2-[4-(3-furyl)-6b,10a-dihydroxy-3a,7,9,9-tetramethyl-6,10-dioxo-2,3,3a,6b,7,8,9,10,10a,11-decahydro-1a*H*,4*H*,6*H*-benzo[*h*][1]benzoxireno[3,2,1a-*de*]isochro-]isochromen-8-yl]acetate, $C_{27}H_{32}O_9$, is a new tetranortriterpenoid which was isolated for the first time from the seeds of *Lansium domesticum* Coor. The four six-membered rings adopt chair, boat, envelope and screw-boat conformations. The furan substituent is planar and is attached equatorially to the lactone ring. $O-H \cdots O$ hydrogen bonds link the molecules into chains along [100] and adjacent chains are interconnected by $C-H \cdots O$ interactions.

Comment

The title compound, (I), is a naturally occurring tetranortriterpene which was isolated for the first time from the seeds of Lansium domesticum Corr., a plant belonging to the Meliaceae family. L. domesticum Corr. has at least four cultivated forms or varieties, namely, langsat, duku, dukulangsat and longkong. This plant is widely cultivated for its edible fruit. In North Borneo, the bark and seeds have been used to treat malaria and fever (Burkill, 1966; Chai et al., 1989). It has also been reported that the Ibans in Sarawak, Malaysia, used the langsat leaves to treat fever (Leaman et al., 1995). As part of our ongoing studies of bioactive compounds from Thai medicinal plants (Chantrapromma et al., 2005; Chantrapromma, Boonnak et al., 2006; Chantrapromma, Fun et al., 2006; Fun et al., 2006), we have undertaken the X-ray crystal structure analysis of (I) in order to establish its molecular structure and relative stereochemistry. Compound (I) was not found to possess antimalarial, antitubercular or antiantitumour activities.



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The structure of (I) contains a four-ring A/B/C/D fused tetranortriterpenoid system which is illustrated in Fig. 1. The



Figure 1

The molecular structure of (I), showing 50% probability displacement ellipsoids and the atom-numbering scheme. The O-H···O hydrogen bond is shown as a dashed line.



The crystal packing of (I), viewed down the a axis. Hydrogen bonds are shown as dashed lines.

A/B ring junction is *cis*-fused. The bond lengths and angles in (I) show normal values (Allen et al., 1987) and are comparable to the corresponding values in a closely related compound, dukunolide F (Nishizawa et al., 1988). The conformations adopted by rings A, B, C and D are chair, boat, envelope with atom C13 in the flap position $[Q = 0.488 (3) \text{ Å}, \theta = 58.4 (4)^{\circ}$

and $\varphi = 244.1 \, (4)^{\circ}$] and screw-boat $[Q = 0.518 \, (2) \, \text{\AA}, \, \theta =$ $66.2 (2)^{\circ}$ and $\varphi = 89.7 (3)^{\circ}$: Cremer & Pople, 1975], respectively. The methyl group at C1 is equatorially attached to ring A. The conformation of the methyl acetate side chain (C21-C23/O6/O7) is (-)-anticlinal with a C1-C2-C21-C22 torsion angle of $-147.41 (19)^{\circ}$. This side chain and two hydroxyl groups are in cis orientations (Fig. 1). The furan ring is planar within ± 0.005 (2) Å and is attached equatorially to the lactone ring D, the torsion angle C14-C15-C17-C20being -86.0 (3)°. The methyl group attached at C14 is *cis* with respect to the furan ring and trans with respect to the two hydroxyl groups. The two hydroxyl groups have β -orientations, similar to those of dukunolide F (Nishizawa et al., 1988) but in contrast with those of dukunolide A (Nishizawa et al., 1985). One of the hydroxyl groups forms an intramolecular hydrogen bond, O8-H8···O1 (Fig. 1), with carbonyl atom O1. In addition, the molecular structure is stabilized by intramolecular $C-H\cdots O$ weak interactions (Table 1).

 $O-H \cdots O$ intermolecular hydrogen bonds (Table 1) link the molecules into chains along [100]. Adjacent chains are interlinked via $C-H \cdots O$ hydrogen bonds (Fig. 2).

Experimental

The seeds (1.6 kg) of L. domesticum Corr., which were collected from Nakhon Si Thammarat province in the southern part of Thailand, were air-dried, powdered and then extracted twice with CH₂Cl₂ at room temperature (five days for each extraction) to yield a yellowgreen viscous liquid (33.2 g). The crude extract was separated by flash column chromatography over silica gel and then eluted with hexane and three further solvents in order of increasing polarity, CH₂Cl₂, EtOAc and CH₃OH, to give 13 fractions (F1-F13). Fraction F5 (9.0 g) was washed with EtOAc followed by recrystallization from EtOAc- CH_2Cl_2 (1:1 v/v), affording colourless single crystals of compound (I) (2.0 g, m.p. 508-510 K).

Crystal data

| $C_{27}H_{32}O_9$ | Z = 4 |
|-------------------------------|---|
| $M_r = 500.53$ | $D_x = 1.385 \text{ Mg m}^{-3}$ |
| Orthorhombic, $P2_12_12_1$ | Mo $K\alpha$ radiation |
| a = 7.9634 (5) Å | $\mu = 0.10 \text{ mm}^{-1}$ |
| b = 14.6700 (9) Å | T = 297 (2) K |
| c = 20.5420 (13) Å | Block, colourless |
| V = 2399.8 (3) Å ³ | $0.50 \times 0.39 \times 0.34 \text{ mm}$ |
| Data collection | |
| Siemens SMART CCD area- | 14158 measured reflections |
| detector diffractometer | 2973 independent reflection |

(i) scans Absorption correction: multi-scan (SADABS; Sheldrick, 1996) $T_{\min} = 0.952, \ T_{\max} = 0.965$

Refinement

1 S

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.042$ $wR(F^2) = 0.120$ S = 1.052973 reflections 332 parameters H-atom parameters constrained

2814 reflections with $I > 2\sigma(I)$ $R_{\rm int} = 0.022$ $\theta_{\rm max} = 27.0^{\circ}$

| $w = 1/[\sigma^2(F_o^2) + (0.0826P)^2]$ |
|--|
| + 0.3239P] |
| where $P = (F_0^2 + 2F_c^2)/3$ |
| $(\Delta/\sigma)_{\rm max} = 0.001$ |
| $\Delta \rho_{\rm max} = 0.30 \text{ e } \text{\AA}^{-3}$ |
| $\Delta \rho_{\rm min} = -0.21 \text{ e } \text{\AA}^{-3}$ |

Table 1Hydrogen-bond geometry (Å, °).

| $D - H \cdot \cdot \cdot A$ | D-H | $H \cdot \cdot \cdot A$ | $D \cdots A$ | $D - \mathbf{H} \cdots A$ |
|-----------------------------|------|-------------------------|--------------|---------------------------|
| O8−H8···O1 | 0.82 | 2.07 | 2.575 (3) | 119 |
| $O9-H9\cdots O6^{i}$ | 0.82 | 2.09 | 2.910 (2) | 174 |
| $C2-H2\cdots O6$ | 0.98 | 2.47 | 2.926 (3) | 108 |
| $C18-H18\cdots O4^{i}$ | 0.93 | 2.58 | 3.452 (3) | 157 |
| C19−H19···O1 ⁱⁱ | 0.93 | 2.55 | 3.096 (4) | 118 |
| $C27 - H27A \cdots O2$ | 0.96 | 2.60 | 2.977 (3) | 104 |
| | | | | |

Symmetry codes: (i) x - 1, y, z; (ii) $-x, y + \frac{1}{2}, -z + \frac{3}{2}$.

H atoms were placed in calculated positions with O–H distances of 0.82 Å and C–H distances in the range 0.93–0.98 Å. The $U_{\rm iso}({\rm H})$ values were set equal to $1.5U_{\rm eq}$ of the carrier atom for hydroxyl and methyl H atoms and $1.2U_{\rm eq}$ for the remaining H atoms. A rotating group model was used for the hydroxyl and methyl groups. In the absence of significant anomalous dispersion effects, Friedel pairs were merged before the final refinement.

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT* (Siemens, 1996); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 1997); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2003).

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